Lesson Plan (Odd Semester) Session 2023-24

Name of the Assistant Professor:Virender Kumar

Class:- B.Sc III (Vth Sem)

Subject:-Chemistry

· Period	Tonico to 1	nemistry
- Ciriou	Topics to be covered	Topic of
		Assignments / Tes
		to be given to the
24 July to 15	Inorganic Chemistry (Chapter D. Markey	students
August	(Chapter 1) Wetal-ligand	
·	bonding in transition metal complexes: Limitation of	•
	valence bond theory, an elementary idea of crystal-field	
	theory, crystal field splitting in octahedral, tetrahedral	
	and square planer complexes, factor affecting crystal-	
	field parameter	
16 August to 31	Inorganic Chemistry (Chapter II) Thermodynamics	
August	and kinetics aspects of metal complexes and factor	
	affecting the stability	
	stashity, hving william Series.	
	substitutions reaction of square planar complexes of Pt[II].	
	refire,	
1 September to	Organic Chemistry (Chapter-I): NMR Spectroscopy-	
5 September	introduction, Principle of nuclear magnetic resonance,	Class Test:
	PMR spectrum, number of signals, peak areas, equivalent	Chapter: Metal-
	and nonequivalent protons are it	ligand bonding in
	and nonequivalent protons positions of signals and	transition metal
	chemical shift, shielding and deshielding of protons,	complexes:
	Proton counting, splitting of signals and coupling	Assignments: NMR
	constants, magnetic equivalence of protons. Discussion	
	of PMR spectra of the molecules: ethyl bromide, n-	
	propyl bromide, isopropyl bromide. Discussion of PMR	
	spectra of the molecules: 1,1-dibromoethane, ethanol,	
	acetaldehyde, ethyl acetate. Discussion of PMR spectra	
	of the molecules toluene beautiful	
	acetophenone Simple problems on PMR spectroscopy	
	for structure determination of organic compounds	
	organic compounds	

16 September -30	Inorganic Chemistry; Chapter 3 (Magnetic properties	
September	of Transition Metal Complexes);	
	Chapter:4 (Electronic Spectra of Transition Metal	
	Complexes)	
01 October-15	Organic Chemistry (Chapter-II): Carbohydrates:	Class Test:
October	Classification and nomenclature of Monosaccharide's .	Organometallic
	D-glucose, Mechanism of osazone formation, Lobry de	Chemistry
	bruyn van Ekenstien rearrangement, Open chain structure	Assignments:
	of glucose and Fructose, interconversion of glucose and	Quantum Mechanics
	fructose, chain lengthening and chain shortening of	
	aldoses, Configurationof monosaccharides, Erythro and	
	threodiastereomers. Conversion of glucose into mannose.	
	Formation of glycosides, Determination of ring size of	
	glucose and fructose. Mechanism of mutarotation.	
	cyclicstructure of D(+)-glucose & D(-) fructose, Structur	
,	es of ribose and deoxyribose.An introduction to di	
	saccharides, sucrose and lactose, Maltose Haworth	
	projection formulae, polysaccharides, Structure of	
	Amylose and amylopectin, Starch and cellulose Organic	
	Chemistry(Chapter-III) Organometallic	
	Compounds: introduction, the Grignard reagents-	
	formation, structure and chemical reactions of Grignard	
	reagent,Organozinc compounds: formation, chemical	
•	reactions, Organo lithium compounds: formation	
16 October-upto	Physical Chemistry	,
exam		

Usadr Kunz