## Lesson Plan (Even Semester) Session 2023-24

Name of the Assistant Professor: - Dr. Richa Rani

Class: - B. Sc 3<sup>rd</sup> Year (6th-Sem.)

Subject: - Physics

Paper – XII: Atomic and Molecular Spectroscopy (PH-602)

| Period           | Topics to be covered (From 05/01/2024)                                                                                                                                                                                                                                                                                                                                                                                                                                                                         | Topic of<br>Assignments<br>/ Tests to be<br>given to the<br>students |
|------------------|----------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|----------------------------------------------------------------------|
| 05 Jan to 15 Jan | Unit – 1: Historical background of atomic spectroscopy                                                                                                                                                                                                                                                                                                                                                                                                                                                         |                                                                      |
|                  | Introduction of early observations, emission and absorption spectra, atomic spectra, wave number, spectrum of Hydrogen atom in Balmer                                                                                                                                                                                                                                                                                                                                                                          |                                                                      |
|                  | series, Bohr atomic model(Bohr's postulates) , spectra of Hydrogen                                                                                                                                                                                                                                                                                                                                                                                                                                             |                                                                      |
|                  | atom, explanation of spectral series in Hydrogen atom, un-quantized<br>states and continuous spectra, spectral series in absorption spectra,<br>effect of nuclear motion on line spectra (correction of finite nuclear<br>mass), variation in Rydberg constant due to finite mass, short comings<br>of Bohr's theory.                                                                                                                                                                                          |                                                                      |
| 16 Jan to 31 Jan | Wilson sommerfeld quantization rule, de-Broglie interpretation of Bohr<br>quantization law, Bohr's corresponding principle, Sommerfeld's<br>extension of Bohr's model, Sommerfeld relativistic correction, Short<br>comings of Bohr-Sommerfeld theory, Vector atom model; space<br>quantization, electron spin, coupling of orbital and spin angular<br>momentum, spectroscopic terms and their notation, quantum numbers<br>associated with vector atom model, transition probability and selection<br>rules. | Unit Test                                                            |
| 01 Feb to 15 Feb | Unit -2: Vector Atom Model (single valance electron)                                                                                                                                                                                                                                                                                                                                                                                                                                                           |                                                                      |
|                  | Orbital magnetic dipole moment (Bohr megnaton), behavior of<br>magnetic dipole in external magnetic field; Larmor's precession and<br>theorem. Penetrating and Non-penetrating orbits, Penetrating orbits on<br>the classical model; Quantum defect, spin orbit interaction energy of<br>the single valance electron, spin orbit interaction for penetrating and<br>non-penetrating orbits. quantum mechanical relativity correction,                                                                          |                                                                      |
| 16 Feb to 29 Feb | Hydrogen fine spectra, Main features of Alkali Spectra and their theoretical interpretation, term series and limits, Rydeburg-Ritze                                                                                                                                                                                                                                                                                                                                                                            | Unit teşt                                                            |

|                         | combination principle, Absorption spectra of Alkali atoms, Observed<br>doublet fine structure in the spectra of alkali metals and its<br>Interpretation, Intensity rules for doublets, comparison of Alkali spectra<br>and Hydrogen spectrum.                                                                                                                            |            |
|-------------------------|--------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|------------|
| 01 March to 15<br>March | UNIT-3: Vector Atom model (two valance electrons)                                                                                                                                                                                                                                                                                                                        | Assignment |
|                         | Essential features of spectra of Alkaline-earth elements, Vector model<br>for two valance electron atom: application of spectra. Coupling                                                                                                                                                                                                                                |            |
|                         | Schemes;LS or Russell - Saunders Coupling Scheme and JJ coupling                                                                                                                                                                                                                                                                                                         |            |
|                         | scheme, Interaction energy in L-S coupling (sp, pd configuration),<br>Lande interval rule, Pauli principal and periodic classification of the<br>elements. Interaction energy in JJ Coupling (sp, pd configuration),<br>equivalent and non-equivalent electrons, Two valance electron system-<br>spectral terms of non-equivalent and equivalent electrons,              |            |
| 16 March to 31<br>March | Comparison of spectral terms in L-S And J-J coupling. Hyperfine structure of spectral lines and its origin; isotope effect, nuclear spin.                                                                                                                                                                                                                                | Unit Test  |
|                         | Unit –4: Atom in External Field<br>Zeeman Effect (normal and Anomalous),Experimental set-up for<br>studying Zeeman effect, Explanation of normal Zeeman effect(classical<br>and quantum mechanical), Explanation of anomalous Zeeman<br>effect(Lande g-factor), Zeeman pattern of D1 and D2 lines of Na atom,<br>Paschen-Back effect of a single valence electron system |            |
| 01 April to 15<br>April | Weak field Stark effect of Hydrogen atom.<br><b>Molecular Physics</b><br>General Considerations, Electronic States of Diatomic Molecules,<br>Rotational Spectra (Far IR and Microwave Region), Vibrational<br>Spectra (IR Region), Rotator Model of Diatomic Molecule, Raman<br>Effect, Electronic Spectra.                                                              | Unit Test  |
| 16 April to 30<br>April | Revision                                                                                                                                                                                                                                                                                                                                                                 |            |

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## Lesson Plan (Even Semester) Session 2023-24

Name of the Assistant Professor: - Mr. Kapil Dev **Class:** - **B. Sc 3<sup>rd</sup> Year (6th-Sem.)** Subject: - Physics Paper – XI: Solid State and Nano Physics (PH-601)

| Period                  | Topics to be covered (From 05/01/2024)                                                                                                                                                                                                                                                    | Topic of<br>Assignments<br>/ Tests to be<br>given to the<br>students |
|-------------------------|-------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|----------------------------------------------------------------------|
| 05 Jan to 15 Jan        | <b>Unit – I Crystal Structure-I:</b> Crystalline and glassy forms, Liquid Crystal, Crystal structure, Periodicity, Lattice and Basis, Crystal translational vectors and axes, Unit Cell and Primitive cell, Winger Seitz primitive cell, Symmetry operations for two dimensional crystal. |                                                                      |
| 16 Jan to 31 Jan        | Bravais lattices in two and three dimension, Crystal Planes and Miller<br>Indices, Interplaner spacing, Crystal Structures of Zinc Sulphide,<br>Sodium Chloride and Diamond.                                                                                                              | Unit Test                                                            |
| 01 Feb to 15 Feb        | <b>Unit-II Crystal Structure-II:</b> X- Ray diffraction, Bragg's law and experimental x-ray diffraction methods, K-space and reciprocal lattice and its physical significance,                                                                                                            |                                                                      |
| 16 Feb to 29 Feb        | Reciprocal lattice vectors, Reciprocal lattice to a simple cubic lattice,<br>B.C.C. and F.C.C.<br><b>Unit-III Superconductivity:</b> Historical introduction, Survey of<br>superconductivity, Superconducting systems, High Tc Super<br>conductors,                                       | Unit test                                                            |
| 01 March to 15<br>March | Isotopic Effect, Critical Magnetic Field, Meissner Effect, London Theory<br>and Pippards' equation, Classification of Superconductors ( Type-I and<br>Type-II).                                                                                                                           |                                                                      |
|                         |                                                                                                                                                                                                                                                                                           | P.T.O                                                                |

| 16 March to 31<br>March | BCS Theory of Superconductivity, Flux quantization, Josephson Effect<br>(AC & DC), Practical Application of Superconductivity and their<br>limitations, Power applications of Superconductors.                                            | Unit Test |
|-------------------------|-------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|-----------|
| 01 April to 15<br>April | <b>Unit- IV- Introduction to Nano Physics:</b> Definition, Length scale,<br>Importance of Nano-scale and technology, History of Nano-<br>Technology, Benefits' and Challenges in molecular manufacturing,<br>Molecular assembler concept. |           |
| 16 April to 30<br>2024  | Understanding advanced Capabilities, Vision and objectives of Nano-<br>technology , Nano-technology in different field, Automobile,<br>Electronics, Nano-Biotechnology, Materials, Medicine.                                              | Unit Test |
| 01 May to 10<br>May 24  | Revision                                                                                                                                                                                                                                  |           |

Jodue 01/02/24